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On the Physical Validity of the LANE-EMDEN Equation in (y; z)-Plane

Lane-Emden equation of index n^1 :

$$\frac{1}{\xi^2} \frac{d}{d\xi} \left(\xi^2 \frac{d\theta}{d\xi} \right) = -\theta^n \,, \tag{1}$$

in (y;z)-plane, takes the form

$$y\frac{dy}{dz} + (2w-1)y + w(w-1) + z^n = 0; (2)$$

where equations connecting (y;z)- with $(\xi;\theta)$ are

$$z = \xi w \theta; \quad w = \frac{2}{n-1} \tag{3}$$

$$y = \frac{dz}{dt} = -\xi^{w+1} \frac{d\theta}{d\xi} - wz; \quad \xi = e^{-t}. \quad (4)$$

It has generally been believed that all parts of the solution-curves which correspond to $\theta \ge 0$, $z \ge 0$ have astrophysical validity. This gives the impression that every point on the y-axis is of interest. But, actually, only points on the positive y-axis have a relevancy.

Equation (3) shows that z vanishes for either $\theta = 0$ or $\xi = 0$. In other words, the neighbourhood of z = 0 in (y;z)-plane maps two different regions in $(\xi;\theta)$ -plane: the regions being the neighbourhood of $\xi = 0$ and that of $\xi = \xi_1$, where solutions of (1) have their first zero at $\xi = \xi_1$. Further equation (4) clears that the former region, in $(\xi;\theta)$ -plane, corresponds to the neighbourhood of the origin y = z = 0 and the latter one to the immediate

neighbourhood of the y-axis. The author has shown recently that in a complete polytrope, whatever be the index of the polytrope, the immediate neighbourhood of the origin is an interfacial region and at the origin solutions for n=0 and n=-1 only are relevant. Hence the point y=z=0 cannot have any relevancy when discussion of solution curves for general values of n is made. When z=0 corresponds to $\xi=\xi_1$, equation (4) shows that y remains positive, since $d\theta/d\xi$ is negative in every part lying in the configuration. Thus we see that the point y=z=0 and the negative y-axis have no physical validity 3.

Zusammenfassung. Es wird bewiesen, dass die Lösungen der Emden-Gleichung nur für y>0 eine physikalische Bedeutung haben.

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Infrared Spectrum of Strontium Hydroxyapatite

The apatites form a group of structurally very closely related substances and hydroxyapatite $\text{Ca}_{10}(\text{PO}_4)_6(\text{OH})_2$, is well known. Relatively little is known about the strontium hydroxyapatite, $\text{Sr}_{10}(\text{PO}_4)_6(\text{OH})_2$. The ionic radii ('crystal' radii of Pauling, in Å) of Sr(1.13) and Ca(0.99) are very close together 1 and strontium can substitute for calcium in the $\text{Ca}_{10}(\text{PO}_4)_6(\text{OH})_2$ lattice. The X-ray powder diffraction patterns confirm the isomorphous substitution of strontium for calcium; the resulting strontium hydroxyapatite crystallizes in hexagonal lattice. Measurements of the unit cell dimensions of strontium hydroxyapatite are given in Table I and are compared with those of hydroxyapatite.

The sample of this investigation was prepared according to the method described by Christensen?. The X-ray powder diffraction pattern resembled that reported by Colling. As far as we know, this is the first IR-spectroscopic study on strontium hydroxyapatite.

The IR-spectrum was recorded on Beckman IR-12. The instrument parameters on the chart legend were: diluent, sample CsI, reference air; concentration, 1.0 mg; phase, solid; gain, 3.0%; ordinate scale, % T; SB/DB energy ratio, 1; speed, 80 cm⁻¹/min.

Under the conditions used, the spectrum resolution was approximately 2-3 cm⁻¹ throughout most of the region. The sample was prepared as a cesium iodide pellet using the following procedure. A small amount of the sample was hand ground in an agate mortar and pestle. 1.0 mg